

MSDS Neutral Alcohol

1. Product and Company Identification

(Page 1 may be used as an emergency safety data sheet)

Trade name : Ethanol (Industrial, Absolute or Anhydrous,

Rum, Light Spirits, Extra Neutral Potable,

Neutral Potable, Rectified Extra Neutral and

High Purity Extra Neutral Potable Alcohol)

Chemical family : Aliphatic Alcohol NIOSH No. : KQ 6300000

Chemical name : Ethanol Hazchem code : 2(S) E; 3(S) E

Synonyms : Ethyl Alcohol, See Trade name UN No. : 1170

2. Composition:

Hazardous components : Ethyl Alcohol $(75.0 - 99.9\%^{\circ}/_{\circ})$

EEC classification : $200 - 578 - 6^{30}$

R Phrases : R11 (Highly Flammable)

3. Hazards Identification:

<u>Main Hazard</u> : Harmful if swallowed or inhaled. Possible aspiration hazard if swallowed (can enter lungs and cause

damage). May be irritating to the skin, eyes and respiratory tract. Over exposure may cause CNS depression.

Chemical abstract No.

Molecular Mass

: 64-17-5

: 46,08

Possible reproductive hazard.

Flammability : Flash Point 12°C -16°C . Extremely flammable liquid (R11). Ignition temperature 425°C.

<u>Chemical Hazard</u>: Ethanol is a flammable liquid whose vapours can form ignitable and explosive mixtures with air at normal

room temperatures. Thus, an aqueous mixture containing 30% ethanol can produce a flammable mixture of vapour and air at 29°C, and even one containing only 5% alcohol can produce a flammable mixture at 62°C.¹ Ethanol reacts vigorously with a wide range of oxidizing materials and other chemicals² .e.g. Disulphuryl Difluoride, Silver Nitrate, Bromine Pentafluoride, Potassium Perchlorate, Nitrosyl Perchlorate, Chromyl Chloride, Chloryl Perchloride, Uranyl Perchlorite, Chromium Trioxide, Fluorine Nitrate, Dioxygen Difluoride, Uranium Hexafluoride, Iodine Heptafluoride, Tetrachlorosilane, Permanganic acid, Nitric acid [the nitric acid fizz reaction used formally for cleaning laboratory glassware should not be used³,5], Hydrogen Peroxide, Peroxodisulphuric acid. Potassium Dioxide, Sodium Peroxide, Potassium Permanganate. Ruthenium (VIII)

Oxide, Platinum, Potassium⁶, Potassium *tert* – Butoxide, Silver Oxide and Sodium⁷.



	water; un-oxidized alcohol is excreted in the urine and expired in the air. ^{8,9}	
Reproductive hazard	: Some evidence of foetotoxicity ²⁶⁻²⁸ and teratogenicity ²⁹ has been observed in experimental animals treated with high doses of ethanol during gestation. Alcohol may induce spontaneous abortions, may impair fertility, may cause harm to the unborn child and may cause harm to breast fed babies. The reproductive hazards have been determined after repeated excessive consumption of ethanol; these effects are not likely to occur through exposure below the Occupational Exposure Limits in the working environment.	
Health effects – eyes	: Moderately irritating. Exposure to liquid, vapours, fumes or mist may cause irritation. Direct contact may cause irritation, redness, pain, corneal inflammation and possible corneal damage.	
<u>Health effects – skin</u>	: Repeated or prolonged contact may result in defatting, redness, pain, itching, inflammation, cracking and possible secondary infection. Repeated skin contact may result in allergic skin reaction in a very small proportion of individuals.	
Health effects — ingestion	: Large doses lead to alcohol poisoning while repeated ingestion can lead to alcoholism. Alcohol abuse and dependence can have a profound effect on work performance and tendency to accidents at work. 11-13 The presence of denaturants, e.g. Methanol, pyridines, and benzene in industrial alcohol greatly increase the toxicity on ingestion. Ethanol drinking is also suspected of increasing the toxic effect of other chemicals encountered in the laboratory and the workplace by inhibition of their metabolism or excretion 14; e.g. 1, 1, 1 — Trichloroethane 15, Xylene, Trichloroethylene and Dimethylformamide 16, Benzene 17 and Lead. 18, 19 May cause harmful central nervous system effects. Effects may include excitation, euphoria, headache, dizziness, drowsiness, blurred vision, fatigue, tremors, convulsions, loss of consciousness, coma, respiratory arrest and death. Severe acute intoxication may cause Hypoglycaemia, Hypothermia and extensor rigidity. Prolonged or frequent contact may result in liver injury.	
Health effects – inhalation	: Intoxicating if continuously inhaled for a long period of time. Occupational Exposure Limits (8-hour reference period) 1000 ppm (1900 mg/m 3). 30	
<u>Carcinogenicity</u>	: Long-term consumption of alcoholic beverages demonstrates an increase in the occurrence of breast cancer and colorectal cancer. Malignant tumours of the oral cavity, Pharynx, Larynx, Oesophagus and Liver is also causally related to the consumption of alcoholic beverages. Some studies have shown an excess incidence of laryngeal cancer over the expected from exposure to synthetic alcohol, with Diethyl Sulphate probably being the causative agent.	
Mutagenicity	: Ethanol has been found to be non- mutagenic in the <i>Salmonella</i> microsome test, ²² but some transient mutagenic changes have been observed in male, but not female, mice treated with rather large doses. ²³⁻² Ethanol is mutagenic in man via its first metabolite, Acetaldehyde. Acetaldehyde induces chromosomal	
4. First – aid Measures:		
Product in eye	: Flush immediately with water or neutral saline solution for at least 15 minutes. Seek medical attention.	
Product on skin	: Remove contaminated clothing and rinse contaminated area with soap and water. If skin irritation persists seek medical attention.	

: Ethanol is rapidly oxidized in the body to acetaldehyde, then to acetate, and finally to carbon dioxide and

Biological Hazard



Product ingested	: If victim is conscious, give 1-3 glasses of water or milk to dilute stomach contents. If spontaneous vomiting occurs, or when vomiting is induced, monitor for breathing difficulty. Do not make an unconscious or semi – conscious person vomit. Keep affected person warm at rest. Get medical attention for substantial ingestions and/or gastrointestinal symptoms.			
Extinguishing media	: Use dry chemical, alcohol foam or carbon dioxide to extinguish fire. Water may be ineffective but should be used to cool fire- exposed containers, structures and to protect personnel. If leak or spill has not ignited, ventilate area and use water spray to disperse gas or vapour and to protect personnel attempting to stop a leak. Use water to dilute spills and to flush them away from sources of ignition. Do not flush down public sewers or other drainage systems.			
Special hazards	: Flammable			
	Flash point : 12 - 16°C			
	Flammability/explosion limits : $3, 3 - 19\%^{v}/_{v}$			
	Dangerous when exposed to heat or flame. Vapours form flammable or explosive mixtures with air at room temperature. Vapour or gas may spread to distant ignition sources and flash back. Run — off to sewer may cause fire or explosion hazard. Containers may explode in heat of fire. Vapours may concentrate in confined areas. Irritating or toxic substances may be emitted upon thermal decomposition.			
Protective clothing	: Exposed fire fighters should wear approved self-contained breathing apparatus with full face mask and full protective equipment.			
6. Accidental Release Measures:				
Personal precautions	: Protective clothing should be worn to prevent excessive skin contact.			
Environmental precautions	: Prevent liquid entering sewers. Do not allow to enter surface waters, storm drains, etc.			
<u>Small spills</u>	: Take immediate steps to stop and contain the spill. Caution should be excised regarding personnel safety and exposure to be spilled material. Eliminate all sources of ignition and wear protective clothing. Absorb small spills onto paper towels and evaporate in a safe place e.g. in a fume hood. Flush the contaminated area with plenty of water.			
<u>Large spills</u>	: Stop leak if you can do it without risk. Contact your local fire department. Eliminate all sources of ignition and static; restrict access to area until completion of clean-up procedure. Wear adequate protective equipment, use self-contained breathing apparatus in confined poorly-ventilated areas. Large quantities should be absorbed on to sand, vermiculite or an equivalent absorbent material and removed to a safe area for disposal. Flush the contaminated area with plenty of water. Incineration is the recommended method of disposal.			



7. Handling and Storage:

Suitable material

: Ethanol is not corrosive to metals and is suitable for food industries. Ethanol may be stored in stainless steel, mild steel or aluminium container. Ethanol may also be stored in HDPE containers.

Handling/

storage precautions

: Ground lines and equipment used during transfer to reduce the possibility of static spark-initiated fire or explosion. Store in approved flammable liquid storage containers. Keep containers tightly closed as this material readily absorbs moisture. Store away from incompatible materials. Store in a cool, dry well-ventilated area away from sparks, flames and other sources of ignition. Eliminate all sources of static electricity. Use non –sparking electrical and ventilation systems. Storage criteria: Flammable Liquid store

8. Exposure Control / Personal Protection:

Occupational exposure limits

:

Country	8 Hour – TWA Hygiene Limit	STEL
US (OSHA)	1900 mg/m³ (1000ppm)	None
US (ACGIH)	1900 mg/m³ (1000ppm)	None
Germany (MAK)*	960 mg/m3 (500ppm)	Peak limit cat. II,1
UK (OES)	1920 mg/m³ (1000ppm)	None
Slovak Republic	960 mg/m3 (500ppm)	1920 mg/m3 (1000ppm)
		(30 min, 4x per shift)
Czech Republic	1000 mg/m ³	3000 mg/m ³

Engineering control measures

: Engineering control methods to reduce hazardous exposures are preferred. General methods include mechanical ventilation (dilution and local exhaust), process or personnel enclosure, control of process conditions and process modification (e.g. substitution of a less hazardous material). Administrative controls and personal protective equipment may also be required. Use a non-sparking, grounded ventilation system separate from other exhaust ventilation systems. Exhaust directly to the outside. Supply sufficient replacement air to make up for air removed by exhaust system.

Personal protection

– respiratory

: If exposure limits are exceeded or if irritation is experienced, an approved respirator for organic vapours is generally acceptable. For high concentrations and for oxygen-deficient atmospheres, use approved air-supplied respirator. Full respiratory protection should be readily available in case of spillage.

Personal protection

<u>– hand</u>

: Rubber (Butyl) or neoprene gloves are recommended.

Personal protection

<u>– eye</u>

: Prevent eye contact with this material. Wear chemical tight safety goggles where eye exposure is reasonably probable. Provide an eyewash station immediately accessible to the work area. Contact lenses should not be worn when working with this chemical.

Personal protection

<u>– skin</u>

: Avoid skin contact. When working with this substance, wear appropriate chemical protective gloves. Depending upon conditions of use, additional protection may be necessary such as face shield, apron, etc.

Other protection

: Provide a safety shower immediately accessible to the work area.



9. Physical and Chemical Properties:

Appearance : Colourless, volatile liquid

Odour : Characteristic pleasant odour

pH : Neutral

Boiling point : 78.2°C - 78.5°C

Melting point : - 130°C to - 112°C

Flash point : $12^{\circ}\text{C} - 16^{\circ}\text{C}$

Flammability : 3, 3 - 19% v/v

Auto-flammability : 363°C

Explosive properties : Vapours can form explosive mixtures with air. All sources of ignition or static must be excluded.

Oxidizing properties : None

Vapour pressure : 59 mm Hg at 20°C

Density : $785.3 \text{ kg/m}^3 - 809 \text{ kg/m}^3 \text{ at } 25^{\circ}\text{C}$

Solubility – water : Miscible with water in all proportions

Solubility – solvent : Miscible with ether, methanol, chloroform and acetone

Solubility – coefficient : 1100 @ 37°C³³

10. Stability and Reactivity:

<u>Condition to avoid</u>: Overheating, flames, sources of ignition or static electricity. Oxidizing agents. Vapour/ air mixtures are

explosive

<u>Incompatible materials</u> : See section 3 (chemical hazards).

Hazardous decomposition products: Incomplete combustion can generate carbon monoxide and carbon dioxide.



11. Toxicological Information:

Acute toxicity : <u>Short-term hazards</u>

 $\begin{tabular}{lll} Acute oral toxicity Ethanol & : LD_{50} \ rat: 7,060 \ mg/kg; \ literature \ value \\ Acute inhalation toxicity Ethanol & : LC_{50} \ rat: 66,000 \ mg/l; \ literature \ value; 4 \ h \\ Acute dermal toxicity Ethanol & : LDLo \ rabbit: 20,000 \ mg/kg; \ literature \ value \\ \end{tabular}$

NOAEL - 2400 mg/kg (2%) - for rats LOAEL - 3600 mg/kg (3%) - for rats

Skin and eye contact : Redness, pain (refer to Section 3 for further information)

Chronic toxicity : Refer to Section 3

Carcinogenicity : Refer to Section 3

Mutagenicity : Refer to Section 3

Neurotoxicity : Refer to Section 3

Reproductive hazards : Refer to Section 3

12. Ecological Information:

 $A quatic toxicity - fish \hspace{1cm} : \hspace{1cm} In \ high \ concentration \ it \ harms \ fish \ and \ plankton; \ LC_{50} \ (fish, 96 \ hours) - 15.3 \ mg/L \ (Pimephales) - 15.3 \ mg/L \ (Pimeph$

promelas)

Aquatic toxicity – daphnia : Threshold for deleterious effects in small crustaceans upwards of 7.800 mg/l;

EC₅₀ (Daphnia, 48 hours) – 5012 mg/L (Ceriodaphnia dubia)

Aquatic toxicity – algae : Toxic threshold concentration: *Pseudomonas putida* upwards of 6.500mg/l, *Scenedesmus*

quadricauda upwards of 5.000mg/l, Microsystis aeruginosa upwards of 1.450ml/L

 IC_{50} (algae, 72 hours) – 275 mg/L

Biodegradability : This product is readily biodegradable. Ethanol is widely recognized as being readily biodegradable in

the environment as it is both a metabolite of and nutrient for microbes. There are no persistent

Bio – accumulation : This product in not expected to bio accumulate through the food chains in the environment. The

very low log KOW of –0.31 is indicative of a low bioaccumulation potential.

Mobility : This product is likely to volatize rapidly into the air because of its high vapour pressure. The product

is poorly absorbed onto soils or sediments. Adsorption coefficient (K_{OC}) solid phase/liquid phase = 1

(highly mobile)

German wgk : 1 (low hazard to water)



13. Disposal Considerations:

Disposal methods : Only under conditions approved by local authorization. See also Section 6.

Disposal of packaging : Empty containers may contain flammable and hazardous residues. Always obey hazard warnings.

14. Transport Information:

UN No : 1170

Substance Identity No : UN 1170

ADR/RID class : 3

ADR/RID item No : 3(b)

ADR/RID hazard identity No : 3

IMDG – shipping name : Ethanol

IMDG – class : 3.2

IMDG – packaging group : II

IMDG – marine pollutant : Not a marine pollutant

IMDG – EMS No : F-E, S-D

IMDG – MFAG table No : 3074

IATA – shipping name : Ethanol Solutions

IATA – class : 3

IATA – subsidiary risk(s) : Flammable liquid

ADNR – class : UN –No.:1170; Class 3, Packaging Group II

UK – description : Not available

UK- emergency action class : Not available

UK – classification : Not available

Tremcard No : 1170

